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# $N$ -(tert-Butoxycarbonylglycyl- $a,\beta$ dehydrophenylalanylglycylphenylalanyl)-4-nitroaniline

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In the crystal structure of the tetrapeptide  $Boc^0$ – $Gly^1$ – $\Delta P$ he<sup>2</sup>– Gly<sup>3</sup>-Phe<sup>4</sup>-p-NA (p-NA is para-nitroaniline),  $C_{33}H_{36}N_6O_8$ , there are two independent molecules differing in conformation in the asymmetric part of the unit cell. All the amino acids in the peptide are linked trans to each other. The torsion angles in the main chain of both molecules are close to the values of the type  $\beta$ -II turn. Two intramolecular and three intermolecular  $N-H\cdots O$  hydrogen bonds stabilize the conformation of each of the molecules.

## Comment

The  $\alpha$ ,  $\beta$ -dehydroamino acid residues have been found to occur naturally in several microbial peptides and antibiotics (Noda et al., 1983; Spatola, 1983). Their presence in peptides confers increased resistance to enzymatic degradation and for this reason many highly active analogues of bioactive peptides have been designed and synthesized (Fisher et al., 1981; Costa et al., 1983; Sharma & Chauhan, 1988). In particular,  $\alpha, \beta$ dehydrophenylalanine ( $\Delta$ Phe) residues exhibit preferential secondary structural features both in the solid state and in solution. Determination of the crystal and molecular structure of many  $\Delta$ Phe-containing peptides has provided evidence that  $\Delta$ Phe is a strong inducer of  $\beta$ -bends (Venkatachalam, 1968) in short sequences with a single  $\Delta$ Phe residue (Główka et al., 1987; Główka, 1988; Aubry et al., 1991) and of  $3_{10}$ -helical structures in long sequences (Rajashankar et al., 1992; Padmanabhan & Singh, 1993; Rajashankar, Ramakumar, Jain & Chauhan, 1995; Rajashankar, Ramakumar, Mal et al., 1995; Jain et al., 1997). Additionally, the final conformation of any  $\Delta$ Phe peptide depends upon the number and position of  $\Delta$ Phe residues as well as the nature of the amino acids flanking them (Rajashankar et al., 1996).

We present here the crystal structure of a tetrapeptide containing one  $\Delta^Z$ Phe between two flexible glycine residues and one phenylalanine, *i.e.*  $Boc^0$ -Gly<sup>1</sup>- $\Delta$ Phe<sup>2</sup>-Gly<sup>3</sup>-Phe<sup>4</sup>-p-NA (p-NA is para-nitroaniline), (I). In the independent part of the unit cell, there are two independent molecules (A and B) differing in conformation (Fig. 1). The bond lengths and angles for the two molecules are the same within  $5\sigma$  (Table 1).

The C<sub> $\alpha$ </sub>=C<sub> $\beta$ </sub> (C8–C9 and C8'–C9') distances for the  $\Delta$ Phe residue in  $A$  and  $B$  agree well with the standard double-bond distance observed in structures containing  $\Delta$ Phe (e.g. Główka, 1988). The conjugation of the aromatic ring with the  $C_{\alpha} = C_{\beta}$ bond is extended to the  $N - C_{\alpha}$  (N2–C8 and N2'–C8') and  $C_{\alpha}$  – C (C8–C16 and C8'–C16') bonds. A shortening of about 0.35 Å for N $-C_\alpha$  and 0.55 Å for  $C_\alpha-C$  is observed with respect to the corresponding bonds in the saturated Phe<sup>4</sup> unit  $(N4 - C19, N4' - C19', C19 - C27$  and  $C19' - C27'$ ). The steric contacts between the side-chain and main-chain atoms of the  $\Delta$ Phe residue are partly relaxed by rearrangement of the bond angles at  $C_{\alpha}$  and  $C_{\beta}$  atoms. For example, the N- $C_{\alpha} = C_{\beta}$  $(N2-C8-C9$  and  $N2'-CS'-C9'$ ) angles are increased from the value of 120° by ca 4°, whereas the  $C_\beta = C_\alpha - C$  (C9–C8– C16 and C9' $-C8$ ' $-C16$ ') angles are reduced by *ca* 2°. Similar effects are observed in Boc-Val- $\Delta$ Phe- $\Delta$ Phe- $\Delta$ Phe-Val-OMe (Jain et al., 1997).



All the amino acids in the peptide are linked trans to each other with a deviation of  $\omega$  from 180° of less than ca 6° [with the exception of 11.8 (2)<sup>o</sup> for  $\omega^4$  for the A molecule]. The torsion angles in the main chains of both molecules and the  $(\Phi/\Psi)$  of Gly<sup>1</sup> and  $\Delta^Z$ Phe<sup>2</sup> approximate the values assigned to the amino acid residues in the corners of a type II  $\beta$ -turn. The intramolecular hydrogen bonds between the amide (N3 and  $N3'$ ) of Gly<sup>3</sup> and the carboxyl (O2 and O2') of the Boc groups are similar to those observed in Boc-Gly- $\Delta$ Phe-Gly-OMe (Główka, 1988). The torsion angles  $\chi^1$  [-7.4 (4) and 9.2 (4)°],  $\chi^{2,1}$  [-27.9 (4) and 29.5 (4)°] and  $\chi^{2,2}$  [155.0 (2) and  $-154.1$  (2)<sup>o</sup>] of the  $\Delta$ Phe residue suggest that the side chains in the  $A$  and  $B$  molecules are planar.

The two Gly residues adopt a conformation known as a polyglycine helix (Walton, 1981), with  $\Phi$  and  $\Psi$  close to 80 and  $-150^{\circ}$ , respectively. The relative inclination of the planes of the two Gly units is 37.5 (3) $\circ$  for A and 31.5 (3) $\circ$  for B. The torsion angles for the Boc group ( $\omega^0$  and  $\Phi^0$ ) in both molecules  $(A \text{ and } B)$  correspond to a *trans-trans* conformation. This makes it possible for O2 (Boc) atoms to take part in an intramolecular N3 $-H3D\cdots$ O2 hydrogen bond (Table 2). The carbonyl oxygen of  $\Delta Phe^2$  takes part in another intramolecular hydrogen bond with the amide group of  $p$ -NA. Those two intramolecular hydrogen bonds stabilize the conformation of both molecules. Each of the molecules is connected to three others by  $N-H \cdots O$  hydrogen bonds of medium strength.





The molecular structure (molecules A and B) of Boc-Gly- $\triangle$ Phe-Gly-Phe-p-NA. Displacement ellipsoids are shown at the 50% probability level. Dashed lines indicate intramolecular hydrogen bonds.

# Experimental

Boc–Gly– $\Delta$ Phe–Gly–Phe– $p$ -NA was synthesized according to the method of Makowski et al. (2001). Crystals suitable for X-ray structure analysis were grown from an ethyl acetate-diethyl ether  $(1:1)$ hexane solution.

## Crystal data



## Data collection



## Refinement

 $\ddot{\phantom{a}}$ 

<sup>4</sup>

 $\overline{1}$ 

 $\epsilon$ 



#### Table 1 Selected geometric parameters  $(\AA, \degree)$ .

 $N2 - C8$  $1.413(3)$  $1.421(3)$  $N2' - C8$  $N4' - C19$  $1.454(3)$  $N4 - C19$  $1.454(3)$  $1.339(4)$  $CS'-C9'$  $C8 - C9$  $1.339(4)$  $C8 - C16$ 1.489 $(4)$  $C8' - C16'$ 1.488 $(4)$  $C19' - C20'$  $C19 - C20$  $1.529(4)$  $1.531(4)$  $C19 - C27$  $1.537(4)$  $C19' - C27$  $1.543(4)$  $C9 - C8 - N2$  $C9' - C8' - N2'$  $124.1(2)$  $124.3(2)$  $C9' - C8' - C16'$  $117.5(2)$  $C9 - C8 - C16$  $118.1(2)$  $N2' - C8' - C16'$  $N2 - C8 - C16$  $117.8(2)$  $117.6(2)$ 

 $= 0.27$  e Å<sup>-3</sup>

#### Table 2 Hydrogen-bonding geometry  $(\mathring{A}, \degree)$ .



Symmetry codes: (i) x, y,  $1 + z$ ; (ii) x,  $y - 1$ ,  $1 + z$ ; (iii)  $1 + x$ ,  $y - 1$ ,  $1 + z$ ; (iv) x, y, z - 1; (v)  $x$ ,  $1 + y$ ,  $z - 1$ ; (vi)  $x - 1$ ,  $1 + y$ ,  $z - 1$ .

The structure is non-centrosymmetric since both molecules have the same configuration  $(S \text{ on } C19)$  and  $C19'$ ). Refinement in the centrosymmetric  $P\overline{1}$  space group was attempted and gave an R value of ca 12%. In the absence of any significant anomalous scatterers, the Friedel equivalents were merged and the absolute configuration set by reference to that of natural phenylalanine. H atoms were treated as riding with  $N-H = 0.90 \text{ Å}$  and  $C-H = 0.96 \text{ Å}.$ 

Data collection: Kuma Diffraction Software (Kuma, 1998); cell refinement: Kuma Diffraction Software: data reduction: Kuma Diffraction Software; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Sheldrick, 1990); software used to prepare material for publication: SHELXL97.

Table 3 Important torsion angles  $(°)$ .

Molecule A	Molecule $B$
$-169.8(2)$	164.0(2)
$-179.1(2)$	$-176.1(2)$
53.9(3)	$-52.7(3)$
177.4(2)	$-178.3(2)$
$-143.8(2)$	149.3(2)
	49.9(3)
	9.2(5)
	29.5(5)
	$-154.2(3)$
	175.3(3)
$-19.2(3)$	29.0(3)
	$-55.6(3)$
	173.9(2)
	130.1(2)
	61.4(3)
	175.0(2)
16.8(3)	20.7(3)
	$-59.3(3)$ $-7.6(4)$ $-27.8(4)$ 155.1(2) $-178.0(2)$ 55.1(2) $-175.9(2)$ $-135.0(2)$ $-91.3(2)$ 168.2(2)

Supplementary data for this paper are available from the IUCr electronic archives (Reference: GD1118). Services for accessing these data are described at the back of the journal.

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